THE ABSOLUTE CONFIGURATION OF THE RHOEADINE ALKALOIDS¹ Maurice Shamma and J.L. Moniot, Department of Chemistry, The Pennsylvania State University, University Park, Pennsylvania, and Wan Kit Chan and Koji Nakanishi, Department of Chemistry, Columbia University, New York, New York (Received in USA 25 August 1971; received in UK for publication 1 October 1971)

The rhoeadine alkaloids are dextrorotatory bases found only in <u>Papaver</u> species (Papaveraceae); three typical examples being (+)-glaudine, (+)-epiglaudine and (+)-oreodine.² The relative stereochemistry at C-1 and C-2 was obtained from nmr coupling constants,³ while the relative stereochemistry at C-14 was derived through nmr and kinetic studies.⁴ It was also established that when glaudine is treated with very dilute acid in methanol, isomerization at C-14 occurs to give epiglaudine. Stronger acid epimerizes irreversibly epiglaudine (B/D trans) to oreodine (B/D cis),⁵ while isomerization again occurs at C-14.



We now wish to report that application of the aromatic chirality method 6 to the above three alkaloids leads to the absolute configurations indicated.

All the rhoeadine alkaloids show characteristic uv maxima at about 290 $(A \rightarrow L_b)$, 240 $(A \rightarrow L_a)$ and 200 nm $(A \rightarrow B)$. As shown in the Table, the 290 and 240 nm uv bands are associated with positive and negative cd extrema respectively. In spite of the rather strong uv absorptions in these two regions (log $\epsilon \approx 3.9$), chromophoric interactions leading to exciton splitting is not observed in the cd spectra due to the relatively long distance between the two chromophores.

In contrast, a Davydov splitting⁷ was observed in the allowed A-+B transition band (log $\epsilon = 5.0$); this was more clearly observed in <u>n</u>-hexane than in ethanol solution due to better transmission. The first Cotton effect was definitely positive in each of the three cases, so that the two chromophores interact as depicted in expressions <u>1</u> and <u>2</u> (positive chirality), thus establishing the absolute configuration. The chiral center at C-14 can be safely regarded as exerting negligible influence on the split Cotton effects.



Since all the rhoeadine alkaloids are strongly dextrorotatory and their relative configurations are known,⁴ it follows that they must all possess the absolute configurations depicted above. The determination of the configurations of the rhoeadines also clearly indicates that acid catalyzed transformation of an epiglaudine to an oreodine analog (trans->cis B/D) proceeds through isomerization at C-1 rather than at C-2.⁸ The following classification can now be made for the rhoeadine bases:^{2,4}

- (a) <u>Glaudine Analogs</u>: Isorhoeadine, papaverrubine A, papaverrubine B, papaverrubine D, alpinine and papaverrubine G.
- (b) <u>Epiglaudine Analogs</u>: Isorhoeagenine, isorhoeagenine-&-D-glucoside, glaucamine, epipapaverrubine B, N-methyl-14-O-desmethylepiporphyroxine, papaverrubine C, alpinigenine and O-methylalpinigenine.
- (c) <u>Oreodine Analogs</u>: Rhoeadine, rhoeagenine, dubirheine, papaverrubine E, oreogenine and papaverrubine F.

Table

CD Data at Room Temperature

Compound	CD Extrema, nm ($\Delta \varepsilon$)	
	Split Cotton Effect ^a	Normal Cotton Effect ^b
(+)-Glaudine	195 (-5.00), 207 (+7.25)	222 (+5.7), 250 (-2.0), 290 (+5.3)
(+)-Epiglaudine	193 (-7.5), 207 (+9.4)	250 (-1.3), 270 (+5.7)
(+)-Oreodine	197 (-10.0), 207 (+12.5)	238 (-3.00), 288 (+3.9)

a. In hexane.

b. In absolute ethanol.

References

- 1. This research was supported by grants NIH HE-12971 (M.S.) and NIH CA-11572 (K.N.).
- For a recent review on the <u>Papaver</u> alkaloids see L. Kuhn, D. Thomas and S. Pfeifer, Wissenschaftliche Zeitschrift der Humboldt-Universität zu Berlin, Math.-Nat. R., <u>29</u>, 1 (1970).
- F. Šantavý, J.L. Kaul, L. Hruban, L. Dolejš, V. Hanuš, K. Bláha and A.D. Cross, Coll. Czech. Chem. Commun., 30, 3479 (1965).
- 4. M. Shamma, J.A. Weiss, S. Pfeifer and H. Döhnert, Chem. Commun., 212 (1968).
- 5. I. Mann, H. Döhnert and S. Pfeifer, Pharmazie, 21, 494 (1966).
- N. Harada, K. Nakanishi and S. Tatsuoka, J. Amer. Chem. Soc., <u>91</u>, 5896 (1969); and
 M. Shamma, J.L. Moniot, W.K. Chan and K. Nakanishi, Tetrahedron Lett., in print.
- S.F. Mason, K. Schofield, R.J. Wells, J.S. Whitehurst and G.W. Vane, Tetrahedron Lett., 137 (1967); and G. Gottarelli, S.F. Mason and G. Torre, J. Chem. Soc. (B), 1349 (1970).
- 8. F. Šantavý, H. Hrbek, Jr., and K. Bláha, Coll. Czech. Chem. Commun., <u>32</u>, 4452 (1967): in this paper assignments of absolute configuration to the rhoeadines were made on the basis of empirical comparisons of ord curves of the rhoeadines with those of the phthalideisoquinoline bases. The conclusions reached were in agreement with the assignments made in the present paper, but such empirical correlations are ambiguous and can at times lead to erroneous results especially when the cd data are as complex as with the rhoeadines.